PHOTOLYSIS OF α-DIAZO-β-KETOSULFONES AND β-KETOSULFONES 1

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(Received 23 January 1967)

A recent communication by De Mayo and co-workers deals with unsuccessful attempts to obtain sulfenes by photolysis of benzylsulfonylmethyl ketones:

$$C_6H_5CH_2-SO_2-CH_2-COR \xrightarrow{hv} C_6H_5-CH=SO_2 + R-C=CH_2$$

No experimental evidence was found for such a Norrish Type II cleavage, involving transfer of a benzylic hydrogen to the carbonyl oxygen. Instead, all products appear to arise from simple cleavage of the carbon-sulfur bond beta to the carbonyl group. The main products are CH₂COR and benzylsulfinic acid.

Sulfenes, as well as photolysis of β -ketosulfones, are subjects of current work in this laboratory 3 also: this letter seems therefore justified.

A carbene II, generated by photolysis of I, might rearrange either to a ketene III (α -ketocarbene behaviour), or to a sulfene IV. From the reaction mixture obtained by irradiation of α -acetyl-p-tolylsulfonyldiazomethane (Ia) in ethanol was isolated ethyl α -(p-tolylsulfonyl)propionate (Va) (20%), the addition product of ethanol to IIIa. Furthermore 17% of acetonyl p-tolyl sulfone (VIa) (reduction product of IIa) and 45% of ethanal (as DNPH) were found. The addition product of ethanol to IV (ethyl α -acetyl-p-tolylmethane-sulfonate) was not obtained. Again, the rearrangement of α -sulfonylcarbene to sulfene seems to be unfavourable 3.

Similarly, photolysis (45 min.) of a-benzoyl-p-tolylsulfonyldiazomethane (Ib) provided benzoylmethyl p-tolyl sulfone (VIb) (33%), acetophenone (20% as DNPH) and more than one mole of ethanal (as DNPH). The presence of ethyl a-phenyl-p-tolylsulfonylacetate (Vb) (addition product of ethanol to IIIb) was indicated by TLC.

The formation of acetophenone and more than one mole of ethanal can be explained by subsequent photolytic decomposition of VIb to a benzoylmethyl free radical, which dehydrogenates additional ethanol. Shorter irradiation (30 min.) raises the yield of VIb to 85%.

It was established in separate experiments that indeed benzoylmethyl p-tolyl sulfone (VIb) gives acetophenone (68%) and ethanal (89%) upon irradiation (1 h) in ethanol. Our results, together with the recent reports in the literature 6 , suggest exclusive cleavage of the sulfur-carbon bond in β -ketosulfones, and not between C=O and CH $_2$ 7 .

Considering the reasonable yields 6a , the photolytic ketone formation might have synthetic value, complementary to the reductive cleavage of β -ketosulfones (and β -ketosulfoxides) with aluminum amalgam 8 . In this respect, the efficient synthesis of β -ketosulfones from esters and dimethyl sulfone with tert-Buok 9 or NaH 8 in DMSO is of interest.

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